

## 2.5 Problems

(2.1) (a)  $1 \text{ yoctomole} = 10^{-24} \text{ mole}$   
 $1 \text{ mole} = 6.022 \times 10^{23} \text{ atoms}$  }  $\Rightarrow 1 \text{ ymole} = 0.6022 \text{ atoms}$

(b)  $1 \text{ nanocentury} = 10^{-9} \text{ centuries}$   
 $1 \text{ century} = 100 \text{ years}$   
 $1 \text{ year} = 365 \text{ days}$   
 $1 \text{ day} = 86400 \text{ seconds}$  }  $\Rightarrow 1 \text{ nCentury} = 3.1536 \text{ seconds}$   
 $\approx \pi \text{ seconds}$

(2.2)  $1 \text{ Tbyte} = 10^{12} \text{ bytes}$   
 $1 \text{ floppy disk (FDD)} = 1.44 \text{ MB} = 1.44 \times 10^6 \text{ bytes}$  }  $\Rightarrow$   
 $1 \text{ Tbyte} = 0.694 \times 10^6 \text{ f.d.s} = 694,000 \text{ f.d.s}$   
 $1 \text{ f.d.} = 3.3 \text{ mm height}$  }  $\Rightarrow$

$1 \text{ Tbyte of f.d.s} = 2290.2 \text{ m} \gg 1 \text{ km} \approx \text{Kingdom Tower}$   
 largest skyscraper in Jeddah

(2.3) Universe  $\approx 10^{80}$  hydrogen atoms  $\equiv 10^{80}$  bits

~~largest number of bits  $2^{10^{78}} + 2^{10^{78}} + \dots + 2^{10^{78}} + 2^{10^{79}}$~~

~~$2^{10^{78}} + 2^{10^{78}} + \dots + 2^{10^{78}} + 2^{10^{79}}$~~

~~$2^{10^{78}} + 2^{10^{78}} + \dots + 2^{10^{78}} + 2^{10^{79}}$~~

~~$2^{10^{78}}$~~

largest number  $(10) = 1 + 2 + 4 + \dots + 2^{10^{78}} + 2^{10^{79}}$   
 $= 2^0 + 2^1 + 2^2 + \dots + 2^{10^{78}} + 2^{10^{79}}$   
 $= 2^{10^{79}} [2^{-10^{79}} + 2^{-10^{78}} + \dots + 2^{-1} + 1]$   
 $\approx 2^{10^{79}} [1 + 1 + \dots + 1 + 1]$

$(2^{10}) \approx 10^3$  so  $10^{3 \times 79}$

OK...

$2^{10^{80}}$	$- 1$
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$$(2.4) \quad g \approx 9.81 \text{ m/s}^2$$

$$g' = \frac{G \cdot 1 \text{ kg}}{1 \text{ m}^2} = \frac{6.673 \times 10^{-11} \text{ m}^3 \text{ kg}^{-1} \text{ s}^{-2} \cdot 1 \text{ kg}}{1 \text{ m}^2} = 6.673 \times 10^{-11} \text{ m/s}^2$$

$$\frac{g}{g'} = 20 \log_{10} \left( \frac{9.81}{6.673 \times 10^{-11}} \right) = \cancel{20 \log_{10} 6.673} - \cancel{20} = \cancel{203.61 \text{ dB}}$$

$$223 \text{ dB}$$

(2.5) Let's find first how many moles is 1 ton of TNT

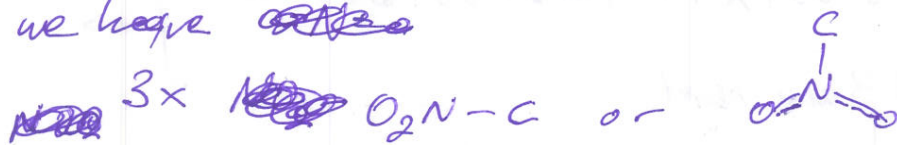
molar mass of TNT:  $227.13 \text{ g} \cdot \text{mol}^{-1}$

$$\# \text{ moles} = \frac{1.000.000}{227.13} = 4402.76 \text{ moles}$$

Bond energies

~~300.000 kJ/mol~~ N=O:  $222 \text{ kJ/mol}$ , N=O:  $607 \text{ kJ/mol}$

in TNT we have ~~300.000~~



result  
 $1.5 \text{ N}=\text{O}$   
 $= 946 \text{ kJ}$

$$\text{Energy}_{\text{ref}} = 4402.76 \times 3 [222 \text{ kJ} + 607 \text{ kJ}]$$

$$\approx 11 \times 10^6 \text{ kJ} = 11 \text{ GJ}$$

Best

$$\text{Energy}_{\text{ref}} = 4402.76 \times 1.5 \times 946 \text{ kJ} = 6 \text{ GJ}$$

$$\text{Thermal energy} = 11 \text{ GJ} - 6 \text{ GJ} = 5 \text{ GJ}$$

$$\text{Actual Number} = 4.184 \text{ GJ (1 Ton)}$$

$$\text{calculation error: } \frac{\cancel{4.184} \text{ GJ} - 4.184}{4.184} \approx 19\%$$



(b)  $10.000 \text{ T of TNT} = 5 \times 10^{15} \text{ J}$   $\xrightarrow{1.6 \times 10^{-19}}$   $3.125 \times 10^{34} \text{ eV}$

In Uranium energy is stored in nucleus weak forces

(c)  $E = m \cdot c^2 \Rightarrow m = \frac{E}{c^2} = \frac{41.84 \times 10^{15} \text{ J}}{9 \times 10^{16} \text{ m}^2/\text{s}^2} = 4.65 \times 10^{-1} \text{ kg} = 4.65 \times 10^2 \text{ g}$   
 $= 465 \text{ g of Uranium.}$

the reverse

$\rightarrow 14.53 \times 10^{28}$  atoms U  $\xrightarrow{235 \text{ g/mole}}$   $569 \text{ g}$  U-235 kind of Do IT with approximate bond energies

(2.6) (a) de Broglie wavelength of baseball

$\lambda = \frac{h}{p}$   $\rightarrow$  thrown baseball  $= 0.230 \text{ kg} \times 41 \text{ m/s} = 9.43 \text{ kg} \cdot \text{m/s}$

so  $\lambda = \frac{6.626 \times 10^{-34} \text{ J} \cdot \text{s}}{9.43 \text{ kg} \cdot \text{m/s}} = 0.7 \times 10^{-34} \frac{\text{kg} \cdot \text{m}^2/\text{s}^2}{\text{kg} \cdot \text{m/s}} = 0.7 \times 10^{-34} \text{ m}$

$\Rightarrow \lambda = 0.7 \times 10^{-34} \text{ m} = 0.7 \times 10^{-24} \text{ \AA}$

(b) de Broglie wavelength of a molecule of nitrogen gas @ room T.  
 we must find  $p$  (momentum)

$[PV = nRT]$  from equipartition Theorem:

3 degrees of freedom  $\rightarrow \frac{3}{2} kT$  thermal energy  $= \frac{1}{2} m v_{\text{ave}}^2$

$\Rightarrow p = m \cdot v_{\text{ave}} = \sqrt{3mkT}$

so  $\lambda = \frac{h}{p} = \frac{h}{\sqrt{3mkT}} = \frac{6.626 \times 10^{-34} \text{ J} \cdot \text{s}}{\sqrt{3 \cdot 14 \text{ g} \times (6.02 \times 10^{23})^{-1} \cdot k \cdot T}} = 0.39 \text{ \AA}$

atomic mass = g/mole

(c) Typical distance between molecules in ~~gas~~ above gas  
 # of molecules in  $V \rightarrow N_A$ , where  $V = 22.4$  litre  
 in STP

$$\frac{V}{N_A} = \text{volume of single molecule.}$$

~~Volume~~  $d_1 d_2 d_3 = \frac{V}{N_A}$

$$\Rightarrow \langle d \rangle^3 = \frac{V}{N_A} \Rightarrow \langle d \rangle = \left( \frac{V}{N_A} \right)^{1/3} = \left( \frac{22.4 \text{ Ltr} \times 10^{-3} \text{ m}^3 \text{ Ltr}^{-1}}{6.022 \times 10^{23}} \right)^{1/3}$$

$$= 33 \text{ \AA}$$

OR sphere approx:  $\frac{4}{3}\pi \langle r \rangle^3 = \frac{V}{N_A}$

$$\Rightarrow \langle r \rangle = \left( \frac{3V}{4\pi N_A} \right)^{1/3} \approx 1.61 \times 33 \text{ \AA}$$

(d)  $\lambda = \frac{h}{\sqrt{3mKT}}$  ~~mean free path~~

$$\Rightarrow \text{I want } \lambda \approx 33 \text{ \AA} \Rightarrow \sqrt{3mKT} = \frac{33 \text{ \AA}}{6.626 \times 10^{-34} \text{ J}\cdot\text{s}}$$

$$\Rightarrow 3mKT = \frac{1}{5} \times 10^{-48} \frac{\text{m}^2 \cdot \text{J}^2}{\text{m}^2}$$

$$\Rightarrow T = \frac{1}{5} \times 10^{-48} \frac{\text{J}^2 \cdot \text{m}^2}{\text{m}^2}$$

$$3 \cdot 0.014 \text{ kg} \cdot 6.022 \times 10^{23} \cdot 1.38 \times 10^{-23} \text{ J/K}$$

$$\Rightarrow T = \frac{0.57}{6.022} \times 10^2 \frac{\text{J}^2 \cdot \text{m}^2}{\text{m}^2 \cdot \text{kg}^2 \cdot \text{J}}$$

$\Rightarrow T = 5.7 \text{ mK}$  wrong calculation but correct approach



(2.7)  $V_M = -\frac{GMm}{r}$



~~Force  $F = \frac{GMm}{r^2}$~~

~~Work  $W = \int F dr = \int \frac{GMm}{r^2} dr = \frac{GMm}{r}$~~

$\int \frac{1}{r^2} dr = -\frac{1}{r}$

$F_M = -\frac{GMm}{r^2} \Rightarrow V_M = \int_0^r F_M dr = +\frac{GMm}{r}$

I want kinetic energy as much as the potential.

$\frac{GMm}{r} = \frac{1}{2}mv^2 \Rightarrow$

$v_{esc} = \sqrt{\frac{2GM}{r}}$

(b)  $c^2 = \frac{2GM}{r} \Rightarrow r_s = \frac{2GM}{c^2}$

Schwarzschild radius horizon

eg earth mass  $= 6 \times 10^{24}$  kg  
 $r_{earth} = \frac{2 \cdot 6.67 \times 10^{-11} \times 6 \times 10^{24}}{9 \times 10^{16}} \text{ m}$   
 $\approx 8.9 \times 10^{-3} \text{ m}$

(c)  $E = M \cdot c^2 \quad \left. \begin{array}{l} \\ \\ \end{array} \right\} \begin{array}{l} M c^2 = \frac{hc}{\lambda} \Rightarrow \lambda = \frac{h}{M \cdot c} \\ E = \frac{h \cdot c}{\lambda} \end{array}$

eg. earth  $M \rightarrow \lambda = \frac{6.626 \times 10^{-34}}{6 \times 10^{24} \times 3 \times 10^8}$

$\rightarrow \lambda = 0.37 \times 10^{-66} \text{ m}$

Compton wavelength  $\leftarrow$  O.M.G too small

(d)  $\lambda = r_s \Rightarrow \frac{h}{M \cdot c} = \frac{2GM}{c^2} \Rightarrow M^2 = \frac{c \cdot h}{2G} \Rightarrow M = \sqrt{\frac{c \cdot h}{2G}} \approx 2.176 \times 10^{-8} \text{ kg}$

PLANK MASS

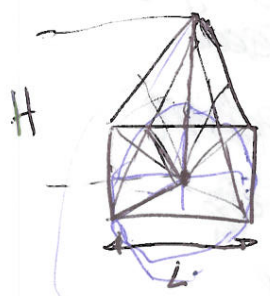
maximum allowed mass for point masses

(e)  $r_s \approx \frac{2 \cdot 6.67 \times 10^{-11} \times 2.176 \times 10^{-8}}{9 \times 10^{16}} \approx 4.05 \times 10^{-35}$  plank size

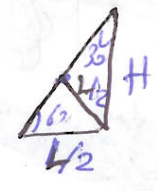
$\leftarrow$  factor of distance

f) Planck energy:  $3.07 \times 10^{28} \text{ eV}$   $E = \frac{1}{2}mv^2$   $T = \frac{1}{v}$   
 g) Planck time:  $1.35 \times 10^{-43} \text{ s}$   $v_{esc} = \sqrt{\frac{2GM}{R}}$

(2.8)



(a)



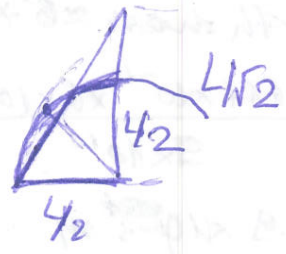
~~tan 60 = H / (L/2)~~



$(\frac{L}{2})^2 + (\frac{L}{2})^2 = (\frac{L}{\sqrt{2}})^2$

$\tan 45 = \frac{H/2}{L/2} \Rightarrow H = L$

$\Rightarrow H = L$



SMART

$\pi \cdot r^2$

$r^2$

$r^2 = \frac{L^2}{4}$

$y = \frac{L^2}{4} - x^2$

$x^2 = \dots$

